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Section 8.1 Assessment page 247 7. Identify the type of atom that generally forms covalent bonds. The majority of covalent bonds form between nonmetallic elements. 8. Describe how the octet rule applies to covalent bonds. Atoms share valence electrons; the shared electrons complete the octet of each atom. 9. Illustrate the formation of single, double, and

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- Covalent Bonding Single Covalent Bonds When only one pair of electrons is shared, such as in a hydrogen molecule, it is a single covalent bond. The shared electron pair is often

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