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13.78 kPa 6. Find the partial pressure of carbon dioxide in a gas mixture with a total pressure of 30.4 kPa if the partial pressures of the other two gases in the mixture are 16.5 kPa and 3.7 kPa. 30.4 kPa 16.5 kPa 3.7 kPa 10.2 kPa Solutions Manual Chemistry: Matter and Change • Chapter 12 237

States of Matter

Notice in particular the curves for NH_4NO_3 and CaCl_2 . The dissolution of ammonium nitrate in water is endothermic ($\Delta H_{\text{soln}} = +25.7 \text{ kJ/mol}$), whereas the dissolution of calcium chloride is exothermic ($\Delta H_{\text{soln}} = -68.2 \text{ kJ/mol}$), yet Figure 13.4.1 shows that the solubility of both compounds increases sharply with increasing temperature. In fact, the magnitudes of the changes in both enthalpy ...

Chapter 13.4: Temperature and Pressure Effects - Chemistry ...

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